ANCIENT ATOMIC THEORY:

to energy also exist for electronic and E, which Attention the basic study of the atomic correspond to theory of **Rutherford**continued improvement **Bohr**; we start from E = T + V. $E = -Ze^2/2r$. By the principle of uncertainty of **Heisenberg** ip = h / λ , mvr = n. \hbar combining the expression of E other gives $r = n^2 \cdot h^2 / (mZe^2)$ according with to Max Planck E = h., $i E = -(mZ^2 . e^4) / (n^2 . h^2 . 2) i E_i = R'.Z^{2}.(1/ni^2)$

Then, depending on non-electronic level (which in the spectrum correspond to Lymann (n = 1), Balmer (n = 2), Paschen (n = 3), Brackett (n = 4) and Pfund (n = 5)) will enjoy an E or another.

$$\Delta E = h. (\nu_i - \nu_j) = \text{R.Z}^2. (\frac{1}{n_i} - \frac{1}{n_j}) \text{ on } n_j > n_i \text{ i R= R'.h.}$$

It is a deeper study of the Hamiltonian and the Schrödinger equation, since it includes the energy of rotation, vibration and electronic + not on ly the subsequent translationals.